## PHOTOCHEMICAL STUDIES WITH O-VINYL DIARYL ETHERS A NEW PHOTO-HETEROCYCLIZATION REACTION ?

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Summary: On photoexcitation the o-vinyl diaryl ethers 10a-d are transformed into o-hydroxy stilbenes 11 which preferentially on direct irradiation are further converted affording the annulated phenanthrenes 12.

The  $6\pi$ -photocyclization is one of the most common excited state reaction which has been successfully performed not only with hexatrienes<sup>1</sup> but also with a great variety of heteroatomcontaining systems of the general structure 1 (X = N-, S, O),<sup>2</sup> which are isoelectronic with the pentadienyl anion.<sup>3</sup> The light induced ring closure of compounds like 1 to the dipolar intermediates  $2^5$  is usually followed either by suprafacial 1,4-hydrogen migration or by oxidation leading to the corresponding five-membered heterocycles 3 and 4, respectively.<sup>2,7</sup>



To the best of our knowledge no analogous ring closure reactions are known for the vinylogous  $8\pi$ -system, which would result in the formation of terminally bridged  $6\pi$ -dipoles (see  $5 \rightarrow 6$ ). Interestingly enough, the reversible formation of intermediates of type 6 has been unambiguously proved by photochemical (or thermal) excitation of epoxy and epimino cyclohexadienes 7.<sup>8</sup>



Here we report on photochemical results obtained with compounds of type 5. In order to avoid unproductive cis/trans isomerizations ("free rotor effect") we started the experiments with o-substituted diphenyl ethers **10a-d** which bear a bridged vinyl group as part of the  $8\pi$ -system.

A straightforward synthesis of the bis-conjugated ethers was accomplished by regioselective ortho-lithiation<sup>9</sup> of diphenyl ether (8) and subsequent reaction with the corresponding cycloalkanones affording the tertiary alcohols 9, which on acid treatment were transformed into the required starting materials 10 (over-all yield 53-79%).<sup>10</sup>



Table 1. UV Data of the Ethers 10a-d (CH<sub>3</sub>CN)

	λ <sub>max</sub> ( ε ) [CH <sub>3</sub> CN]							
10 a	288 (2500), 253 (sh, 13800), 220 (21300)							
10 b	277 (1400), 220 (15000)							
10 c	278 (3300), 241 (sh, 12500), 221 (19500)							
10 d	276 (3300), 235 (sh, 14400), 220 (18800)							

The UV data of **10b-d** are rather similar with long wavelength maxima at approximately 277 nm; only **10a** shows a red shift by 10-12 nm.

Direct irradiation of ca.  $10^{-3}$  M solutions of **10a-d** in ether or methanol has been performed with a high pressure mercury lamp using a vycor filter ( $\lambda > ca. 240$  nm; despite absorption bands at 270-280 nm, solidex filtered light ( $\lambda > ca. 280$  nm] gave extremely slow reactions). Depending on the photolysis conditions one or two products were formed in each case (see Table 2). As exemplified with **10a**, irradiation of 40 mg for 10 min afforded only one compound, namely the cyclopenteno phenanthrene **12a** in almost quantitative yield (run 1). Under the same conditions 100 mg of **10a** were transformed into a 5:1 mixture of **12a** and the phenol derivative **11a** (run 2). On the other hand, an almost selective formation of the latter compound was observed by using acetone as solvent (run 3). Obviously the stilbene **11a** is the primary photoproduct which is converted preferentially from the singlet excited state into the phenonthrene **12a**.<sup>11</sup>



Compared to **10a** the photoreactivity of the *o*-cyclohexenyl diaryl ether **10b** is considerably reduced. Nevertheless after 3 to 9 fold irradiation times the analogous products, **11b** and **12b**, were formed, yet in slightly smaller yields with the phenol **11b** dominating on both direct and sensitized photolysis (run 4 and 5; in addition to **11b** an isomer is isolated to a minor extent, in which the cyclohexene double bond is only conjugated to the phenol ring). In the case of the cycloheptene and cyclooctene derivatives **10c**,**d** the phenonthrenes **12** were the sole monomeric products identified by spectroscopic analysis before and after work-up.

Run		mg	Conditions Solvent (100 ml)	Filter	Time (min)	Conversion	Yield 11	(%) <sup>a)</sup> 12
1	10 a	40	Et <sub>2</sub> O	Vycor	10	100	-	95
2	10 a	100	MeOH	Vycor	10	88	11(12)	50(57)
3	10 a	50	acetone	Solidex	10	95	84(88)	<2
4	10 b	100	MeOH	Vycor	90	80	44(55)	20(25)
5	10 b	100	acetone	Solidex	30	79	58(73) <sup>b)</sup>	5(7)
6	10 c	100	MeOH	Vycor	30	50	-	20(40)
7	10 c	50	acetone	Solidex	30	65	-	trace
8	10 d	100	MeOH	Vycor	60	55	-	40(72)
9	10 d	50	acetone	Solidex	30	36	<u> </u>	<3

Table 2. Photoreactions of 10a-d (150 W lamp, Hanau TQ 150)

a) Values in parentheses are refering to conversion b) 9 (11) % of the isomer (see text)

The structure of the photoproducts 11a,b and 12a-d (each isolated in pure form by column and medium pressure chromatography) has been clearly attributed on the basis of the analytical data 10 12a12 and 12b13 are known compounds.

A possible - however speculative - mechanism for the formation of the phenol system 11 consists of four steps: (i) 8n-cyclization of the electronically excited ether 10 leading to the carbonyl vlide 15,14 (ii) photoinduced (or thermal) cleavage of the C-O bond leading to the carbene 16.15 which (iii) undergoes rapid 1.2-H-shift to the o-quinone methide 17;16 (iv) 1,5-H migration with formation of the observed phenol derivatives 11.17 With the exception of (i) all steps have literature precedents.

Surprisingly, no evidence has been obtained for the photochemical  $6\pi$ -process involving 13 or 14 which would give rise to dibenzofuran derivatives after subsequent H-shift or oxidation<sup>2,18</sup> (experiments in the presence of O2 or J2 were unsuccessful). This observation, together with the





12



18



11

hν

failure of getting any indication for the formation of dibenzooxepin derivatives prompted further studies on the photochemistry of *o*-vinyl diaryl ethers and related compounds. As shown in the succeeding communication, a different mechanism has been developed on the basis of results with specifically substituted compounds.

The photolability of **11** is easily explained by the well established stilbene-dihydrophenanthrene cyclization<sup>11</sup> producing the alcohol **18** which is rapidly transformed under loss of water into the annulated phenanthrenes **12**. In accordance with other results the ring closure process takes place preferentially *via* the singlet excited state.

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